PTO/SB/17 (10-03)

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for FY 2004 Effective 10/01/2003. Patent fees are subject to annual revision.

Applicant claims small entity status. See 37 CFR 1.27

TOTAL AMOUNT OF PAYMENT

(\$)	320

Complete if Known				
Application Number	09/699,723			
Filing Date	October 30, 2000			
First Named Inventor	Goldstein, Joel			
Examiner Name	Riddick, Marie L.			
Art Unit	1713			
Attorney Docket No.	06076 USA			

METHOD OF PAYMENT (check all that apply)	FEE CALCULATION (continued)					
Check Credit card Money Other None	3. ADDITIONAL FEES					
✓ Deposit Account:	Large Entity Small Entity					
Denosit	Fee Code	Fee (\$)	Fee Code	Fee (\$)	Fee Description	5 D-:-
Account Number 01-0493	1051	130	2051	• •	Surcharge - late filing fee or oath	Fee Paid
Deposit Air Products and Chamicala Inc.	1052	50	2052		Surcharge - late provisional filing fee or	
Account Name Air Products and Chemicals, Inc.					cover sheet	
The Director is authorized to: (check all that apply)	1053 1812	130	1053		Non-English specification For filing a request for ex parte reexamination	
☐ Charge fee(s) indicated below ☐ Credit any overpayments	1804	2,520 920*	1812 1804	-	• • • • • • • • • • • • • • • • • • • •	
Charge any additional fee(s) or any underpayment of fee(s)	1004	920	1004	920	Requesting publication of SIR prior to Examiner action	
Charge fee(s) indicated below, except for the filing fee	1805	1,840*	1805	1,840*	Requesting publication of SIR after	
to the above-identified deposit account.	4054	440	2054		Examiner action	
FEE CALCULATION	1251 1252	110 420	2251 2252	55 210	Extension for reply within first month Extension for reply within second month	
1. BASIC FILING FEE	1252	950	2252		• •	
Large Entity Small Entity Fee Fee Fee Fee Description Fee Paid	1254		2254	740	Extension for reply within third month	
Code (\$) Code (\$)					Extension for reply within fourth month	
1001 770 2001 385 Utility filing fee	1255		2255		Extension for reply within fifth month	
1002 340 2002 170 Design filing fee	1401	330	2401		Notice of Appeal	320
1003 530 2003 265 . Plant filing fee	1402	330	2402		Filing a brief in support of an appeal	320
1004 770 2004 385 Reissue filing fee	1403	290	2403		Request for oral hearing	
1005 160 2005 80 Provisional filing fee	1451		1451		Petition to institute a public use proceeding	
SUBTOTAL (1) (\$) 0	1452	110	2452		Petition to revive - unavoidable	
2. EXTRA CLAIM FEES FOR UTILITY AND REISSUE	1453		2453		Petition to revive - unintentional	
Fee from Ext <u>ra Claims below</u> Fee Paid	1501 1502	1,330 480	2501 2502		Utility issue fee (or reissue)	
Total Claims	1502	640	2502		Design issue fee Plant issue fee	
Independent Claims - 3** = X 0	1460	130	1460		Petitions to the Commissioner	
Multiple Dependent	1807	50	1807		Processing fee under 37 CFR 1.17(q)	
Large Entity Small Entity	1806	180	1806		Submission of Information Disclosure Strnt	
Fee Fee Fee Fee <u>Fee Description</u> Code (\$) Code (\$)					Recording each patent assignment per	
1202 18 2202 9 Claims in excess of 20	8021	40	8021		property (times number of properties)	
1201 86 2201 43 Independent claims in excess of 3	1809	770	2809	385	Filing a submission after final rejection (37 CFR 1.129(a))	
1203 290 2203 145 Multiple dependent claim, if not paid	1810	770	2810	385	For each additional invention to be	
1204 86 2204 43 ** Reissue independent claims over original patent	1801	770	2801	385	examined (37 CFR 1.129(b)) Request for Continued Examination (RCE)	
1205 18 2205 9 ** Reissue claims in excess of 20 and over original patent	1802	900	1802	900	• • • • • • • • • • • • • • • • • • • •	
SUBTOTAL (2) (\$) 0	Other fee (specify)					
**or number previously paid, if greater; For Reissues, see above	*Reduced by Basic Filing Fee Paid SUBTOTAL (3) (\$) 320				320	

SUBMITTED BY				(Complete	(if applicable))
Name (Print/Type)	Mary E. Bongiomo	Registration No. (Attorney/Agent)	36091	Telephone	610-481-8820
Signature	Mary & Bongror	no		Date	04/05/04

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DOCKET NO.: 06076 USA

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

IN RE APPLI-

CATION OF :

Goldstein, et al.

: CONFIRMATION NO.: 6932

SERIAL NO.

09/699.723

: GRP. ART UNIT:

1713

FILED

October 30, 2000

: EXAMINER: Reddick, Marie L.

FOR

REDUCED FORMALDEHYDE NONWOVEN BINDERS WHICH

CONTAIN POLYMERIZED UNITS OF N-METHYLOLACRYLAMIDE

CUSTOMER NO.:

23543

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Sir:

APPEAL BRIEF UNDER 37 CFR 1.192(b)

This appeal is from the final rejection mailed on May 21, 2003.

REAL PARTY IN INTEREST

Air Products Polymers, L.P. is the real party in interest in the appeal. The assignment has been recorded at Reel/Frame 012762/0076.

RELATED APPEALS AND INTERFERENCES

There are no related appeals or interferences.

STATUS OF CLAIMS

Claims 1-7 are pending and are being appealed. Claims 1-7 were rejected in a final office action dated May 21, 2003, and an advisory action mailed on February 3, 2004.

STATUS OF THE AMENDMENTS

All amendements have been entered.

SUMMARY OF INVENTION

The invention relates to use of a special class of reducing agents that is part of a redox catalyst system for the emulsion polymerization of vinyl acetate, N-methylolacrylamide (NMA), and, optionally, another monomer. The reducing agents have the formula

$$\begin{array}{c|c}
O & R_1 \\
S & R_2 \\
R_3 & R_3
\end{array}$$

wherein M is a hydrogen atom, an ammonium atom or a monovalent metal ion, R_1 is OH or NR_4R_5 wherein R_4 and R_5 each are H or C_1 - C_6 alkyl; R_2 is H or an alkyl, alkenyl, cycloalkyl or aryl and R_3 is CO_2M . The preferred reducing agent is the glycolic acid adduct of sodium sulfite which is sold under the trademark Bruggolite FF-6. (page 6, lines 16-18, of the specification) One of the unexpected benefits of using these reducing agents as part of the redox catalyst system is that the polymer emulsion has reduced free formaldehyde content. (page 3, lines 1-14; and page 5, line 22 to page 6, line 18, of the specification)

ISSUES

There is one rejection presented in the final office action. Claims 1-7 are rejected under 35 U.S.C. 103(a) as being unpatentable over Berghofer et al. (US 6,211,400) in combination with Applicants' disclosure.

GROUPING OF CLAIMS

Claims 1 through 7 are to be considered as one group.

ARGUMENT

The Examiner maintained the rejection of Claims 1-7 under 35 USC 103(a) as being unpatentable over Berghofer et al (US 5,211,400 B1) in combination with Applicants' own disclosure. The Examiner stated at page 3 first full sentence of the final rejection (paper number 15):

"It is tenable that the sulfinic acid derivatives (reducing agents) used in the emulsion polymerization technique of Berghofer et al would generate a final polymer emulsion governed by a reduction in formaldehyde since the vinyl acetate based polymer emulsion of Berghofer et al, as modified, is essentially the same as the claimed vinyl acetate based polymer emulsion and there is nothing iron clad on this record diffusing this issue."

In the office action mailed on May 21, 2003 (paper number 13), at page 4, first paragraph, the Examiner referred to the Background of the Invention at page 1, lines 20-21 and page 2, lines 1-15 of Applicants' disclosure as support for modifying the polymer disclosed in Run 11 of Berghofer et al. to include NMA as a co-monomer. However, Berghofer et al. does not disclose or suggest using a crosslinking monomer in the polymers disclosed therein and specifically does not disclose or suggest using NMA as a crosslinking monomer. Use of a crosslinking monomer, such as NMA, produces a distinctly different polymer than the polymers disclosed by Berghofer et al.

It should also be noted that page 1, lines 12-14, of the instant specification, reads: "Reduction of formaldehyde in vinyl acetate based emulsions has been achieved by using less favored reducing agents to the formaldehyde sulfoxylates, or by reducing the level of N-methylol acrylamide. ..."

Since Berghofer et al. teach that the sulfinic acid derivatives disclosed therein have a reducing action similar to formaldehyde sulfoxylates but do not eliminate formaldehyde before, during, and after use, it would not be expected that use of the sulfinic acid derivatives in producing polymers from vinyl acetate, NMA, and optionally another monomer, would result in a reduction in residual formaldehyde.

As the Examiner indicated, the reference is evaluated, as a whole, for what it fairly teaches. It teaches, as a whole, that the novel compounds disclosed therein can be used as reducing agents and will not eliminate formaldehyde before, during and after use. Berghofer et al. do not disclose or suggest polymerization in which residual formaldehyde is formed, such as when NMA is used as a co-monomer.

The ability to use NMA in the polymer and reduce the amount of formaldehyde that would typically be present in the resulting emulsion polymer is the value of the reducing agents of this invention. The formaldehyde-free reducing agent of this invention

unexpectedly reduces the amount of formaldehyde in the final polymer emulsion product derived from vinyl acetate vinyl acetate, NMA and optionally other monomers, compared to other formaldehyde-free reducing agents. See Example 2 of this specification in which polymers of vinyl acetate, ethylene, and NMA were produced using either ascorbic acid (sodium erythorbate) or Bruggolite FF 6 as reducing agent. Use of sodium erythorbate resulted in no additional formaldehyde being formed during the polymerization process. However, unexpectedly, use of Bruggolite FF 6, in the same process, resulted in a substantial reduction of formaldehyde. A summary of the 3 different runs reported in Example 2 is presented below:

	Amount of Formaldehyde (ppm) using One of the following Reducing Agents:				
Vinyl Acetate-Ethylene-NMA Polymer	Sodium Erythorbate	Bruggolite FF 6			
A (Tg = -14 °C)	27.2	3.3			
B (Tg = 10 °C)	57.1	8.6			
C (Tg = 0 °C)	47.5	6.8			

The above described reduction in formaldehyde is not reported by Berghofer et al. In fact, Berghofer et al state, at col. 4, lines 22-24, that the reducing compounds disclosed therein have a reducing action comparable to formaldehyde sulfoxylate; but they do not eliminate formaldehyde before, during, or after use.

The above data rebut a prima facie obviousness rejection based on Berghofer et al together with what is well known and admitted in the Background of this Invention.

It is therefore submitted that the claimed invention would not have been obvious based on Berghofer et al together with what is well known and admitted in the Background of this Invention. Berghofer et al do not teach preparation of polymers containing NMA, and do not teach or suggest that formaldehyde would be lowered during the preparation of polymer emulsions in which reducing agents described therein are used. In addition, the data in this case rebut a prima facie obviousness rejection based on Berghofer et al together with what is well known and admitted in the Background of this Invention.

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Reversal of the rejection is requested.

Respectfully submitted,

Mary & Bongums Mary E. Bongiorno Agent for Applicants Registration No. 36,091

7201 Hamilton Boulevard Allentown, PA 18195-1501 (610) 481-8820

APPENDIX

Claims Involved in the Appeal

1. In a vinyl acetate based polymer emulsion formed by the emulsion polymerization of vinyl acetate and N-methylolacrylamide, optionally other monomers, in the presence of a stabilizing system and a redox catalyst system comprised of an oxidizing agent and a reducing agent, the improvement for reducing formaldehyde emissions in the resulting vinyl acetate based polymer emulsion, which comprises:

forming said vinyl acetate based polymer emulsion utilizing as the reducing component of the redox catalyst system a reducing agent of the formula:

$$\begin{array}{c|c}
O & R_1 \\
S & R_2 \\
R_3 & R_3
\end{array}$$

where M is a hydrogen atom, an ammonium atom or a monovalent metal ion, R_1 is OH or NR_4R_5 wherein R_4 and R_5 each are H or C_1 - C_6 alkyl; R_2 is H or an alkyl, alkenyl, cycloalkyl or aryl and R_3 is CO_2M .

- 2. The vinyl acetate based polymer emulsion of Claim 1 in which the vinyl acetate based polymer comprises polymerized units of ethylene in an amount of from about 10 to 40% by weight of the polymer.
- 3. The vinyl acetate based polymer emulsion of Claim 2 wherein the N-methylolacrylamide is present in an amount of from about 0.5 to 10% by weight of the polymer.
- 4. The vinyl acetate based polymer emulsion of Claim 3 wherein the reducing agent represented by the formula is selected from the group consisting of: 2-hydroxyphenyl hydroxymethyl sulfinic acid-sodium salt; 4-methoxyphenyl hydroxymethyl sulfinic acid-sodium salt; 2-hydroxy-2-sulfinato acetic acid-disodium salt; 2-hydroxy-2-sulfinato acetic acid-zinc

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salt; 2-hydroxy-2-sulfinato propionate-disodium salt; ethyl 2-hydroxy-2-sulfinato propionate-sodium salt.

- 5. The vinyl acetate based polymer emulsion of Claim 3 wherein the vinyl acetate based polymer emulsion is formed using a redox catalyst system of hydrophobic hydroperoxide and the glycolic acid adduct of sodium sulfite.
 - 6. The vinyl acetate based polymer emulsion of Claim 3 wherein M is sodium or zinc.
 - 7. The vinyl acetate based polymer emulsion of Claim 3 wherein R_1 is OH.